

TETRAHEDRON LETTERS

Tetrahedron Letters 44 (2003) 6509-6511

## Regiocontrolled synthesis of highly-functionalized fused imidazoles: a novel synthesis of second generation LFA-1 inhibitors

Rogelio P. Frutos\* and Michael Johnson

Department of Chemical Development, Boehringer Ingelheim Pharmaceuticals, Inc., 900 Ridgebury Rd./PO Box 368, Ridgefield, CT 06877-0368, USA

Received 16 April 2003; revised 15 June 2003; accepted 15 June 2003

**Abstract**—A new and reliable route to a new class of LFA-1 inhibitors such as (2) has been developed. A key aspect of this route is the transformation of amino amide 5 into iodide 3 in four steps. Iodide 3 is a key advanced intermediate used in the synthesis of all second-generation 1H-imidazo[1,2- $\alpha$ ]imidazol-2-one LFA-1 inhibitors. © 2003 Elsevier Ltd. All rights reserved.

Lymphocytes are cells of the immune system that must balance their dual role of surveillance and responsiveness as they patrol the body in search of foreign antigens. In their surveillance role, they must circulate freely in the blood and lymph in a non-adherent (unactivated) state. In the presence of a foreign antigen, however, these cells must become activated quickly, cross membrane walls, congregate at the site of infection and adhere to and destroy the invading cells bearing the foreign antigen. The interactions between cells that direct lymphocyte activation, migration and localization are mediated by cell-surface molecules known as adhesion receptors.1 LFA-1 (lymphocyte function-related antigen) belongs to the integrin family of receptors associated with the role of regulating adhesion and migration of lymphocytes, and the counter receptor to LFA-1 on target cells is recognized to be ICAM-1 (intercellular cell adhesion molecule).

Blocking the protein–protein interaction of cell adhesion molecules such as LFA-1 to ICAM-1 has the potential to benefit patients suffering from immune disorders. Accordingly, a program to find small molecule inhibitors of LFA-1 was initiated by our discovery team and resulted in the finding of series of hydantoins inhibitors such as BIRT377 (1).<sup>2</sup> Further SAR studies led to a series of structurally related 1*H*-imidazo[1,2-α]imidazol-2-ones such as compound 2 with improved pharmacological properties (Fig. 1).<sup>3</sup>

Keywords: imidazoles; LFA-1 inhibitors; vinyl iodides.

The selection 2 and related compounds for further pre-clinical studies created the need to develop a safe, robust, reliable and scalable process suitable for the synthesis of large amounts of these compounds. The development of such process presented us with a number of challenges, as certain aspects of the original discovery route were not ideal for scale-up due to the use of potentially dangerous azide reagents and the extensive use of silica gel chromatography. In addition, a key step of the discovery route involved the direct iodination of 4 to afford 3, and this reaction was not regioselective and resulted in the formation of a mixture of regioisomers that had to be separated by chromatography. 4,19

We focused our attention on the highly functionalized vinyl iodide 3 (Scheme 1), as this compound is a key precursor to all second generation 1H-imidazo[1,2- $\alpha$ ]imidazol-2-one LFA-1 inhibitors such as 2. Therefore, a safer, efficient and scalable route to iodide 3 would not only make it possible to synthesize large

Figure 1.

<sup>\*</sup> Corresponding author. Tel.: 203-798-4681; fax: 203-791-6130; e-mail: rfrutos@rdg.boehringer-ingelheim.com

Scheme 1. Retrosynthesis analysis of iodides 3 and 2.

amounts of compounds such as 2, but it would also facilitate the synthesis of analogs and expedite SAR studies. Key objectives of our new synthesis of 3 (Scheme 1) would be to eliminate the use of azide reagents and to replace the non-selective iodination of 4. Accordingly, our proposed synthesis of 3 is shown in Scheme 1. Iodide 3 would be synthesized from intermediate 5, which has the same oxidation state as 3. Intermediate 5 would be prepared from intermediate urea 6, and 6 would come from amino amide 7, for which a well-established synthetic route from D-alanine already exists.<sup>5</sup>

Our first generation synthesis of intermediate 5 is shown in Scheme 2. Treatment of 7 with commercially available ethyl isocyanoacetate afforded urea 6 in excellent yield as a crystalline compound. Dehydration of urea 6 upon treatment with Ph<sub>3</sub>P·CCl<sub>4</sub>/Et<sub>3</sub>N<sup>6</sup> gave intermediate 8 in good yield, presumably through a transient carbodiimide that cyclizes spontaneously. Subsequent hydrolysis of 8 with LiOH gave carboxylic acid 9, which was converted to its acyl chloride with oxalyl chloride and triethylamine and cyclized under the reaction conditions to afford 5 in 57% yield.

Further optimization led to a more streamlined synthesis of 5 (Scheme 3). We were able to combine the

**Scheme 2.** Reagents and conditions: (a) ethyl isocyanoacetate,  $CH_2Cl_2$ , rt, 2.5 h; (b)  $Ph_3P$ ,  $CCl_4$ ,  $Et_3N$ , rt, 12 h (81%); (c) 1 M LiOH, THF, rt, 1 h; (d)  $(COCl)_2$ ,  $Et_3N$ , THF,  $-20^{\circ}C$  (57% for two steps).

Scheme 3. Reagents and conditions: (a) i. ethyl isocyanoacetate, CH<sub>2</sub>Cl<sub>2</sub>, rt, 2.5 h; ii. Ph<sub>3</sub>P, CCl<sub>4</sub>, Et<sub>3</sub>N, rt, 12 h (81%); (b) Me<sub>3</sub>Al, Ph<sub>3</sub>PO, toluene, 25°C (91%).

two-step synthesis of **8** into a one-pot procedure without isolation of **6**, and accomplished the direct synthesis of **5** from **8** without going through intermediate **9**. Accordingly, treatment of **7** with ethyl isocyanoacetate followed by in situ dehydration and cyclization of the intermediate urea (**6**) afforded **8** in 81% yield. Heating **8** in the presence of acids or bases failed to produce **5**, but intermediate **8** could be converted to **5** using a modification of Weinreb's procedure for the conversion of esters to amides.<sup>8,9</sup>

With a reliable source of 5 now available, we turned our attention to the transformation of 5 into iodide 3. The direct synthesis of vinyl halides such as chlorides and bromides from amides and ketones with reagents such as POCl<sub>3</sub> or POBr<sub>3</sub><sup>10</sup> is well known, <sup>11</sup> but the synthesis of vinyl iodides from carbonyl compounds is far less common, and we could not find a literature precedent for the direct or indirect transformation of amides into vinyl iodides. There exist a number of methods for the indirect synthesis of vinyl iodides from ketones through intermediates such as hydrazones,12 vinyl trimethyltin compounds,13 triflates14 and vinyl phosphates, 15 and we decided to pursue the use of a vinyl phosphate intermediate due to the reported higher stability of such compounds relative to similar triflates.16

Phosphate 10 was easily prepared from 5 in good yield using standard conditions (KN(TMS)<sub>2</sub>, (EtO)<sub>2</sub>POCl, THF), and could be purified by filtration through a short plug of silica gel and stored for several weeks without noticeable decomposition (Scheme 4). Phosphate 10 was then converted to 3 in 75% yield upon treatment with NaI/TMSCl according to a modification of the procedure described by Wiemer and co-workers<sup>15</sup> in their synthesis of vinyl iodides from ketone-derived vinyl phosphates. Interestingly, the formation of 3 was faster and more reproducible when 0.7 to 1 equiv. of water were added to the reaction mixture. There is very little known about the mechanism for this transformation, but it is very likely that at least in our case, the presence of hydrogen iodide generated by the TMSCI/ NaI/water<sup>17</sup> system is crucial for the reaction.<sup>18</sup>

With a reliable and convenient route to intermediate 3 at hand, the preparation of second generation 1*H*-imi-

**Scheme 4.** Regents and conditions: (a) KN(TMS)<sub>2</sub>, (EtO)<sub>2</sub>POCl, THF, -20°C (92%); (b) NaI, TMSCl, H<sub>2</sub>O, CH<sub>2</sub>Cl<sub>2</sub> (75%).

Scheme 5. Reagents and conditions: (a) c-PentMgCl, SO<sub>2</sub>, NCS, R<sub>2</sub>NH, THF (89%); (b) ArB(OH)<sub>2</sub>, (dppf)PdCl<sub>2</sub>, K<sub>3</sub>PO<sub>4</sub>, DME (80–85%).

dazo[1,2- $\alpha$ ]imidazol-2-one LFA-1 inhibitors such as **2** in large quantities was accomplished using the procedure described by Wu and co-workers<sup>4,19</sup> (Scheme 5).

In conclusion, we have developed a new, reliable, regiocontrolled and safe route to a new class of LFA-1 inhibitors such as **2**. The novel aspect of this route is the transformation of amino amide **7** into iodide **3** in four steps. Development of a reliable synthesis of **3** is important because **3** is a key advanced intermediate used in the synthesis of second-generation 1H-imidazo $[1,2-\alpha]$ imidazol-2-one LFA-1 inhibitors.

## Acknowledgements

The authors wish to thank Dr. V. Farina, Dr. Chris Senanayake and Dr. L. Nummy for insightful suggestions.

## References

- For a detailed discussion of adhesion receptors of the immune system, see: Springer, T. A. Nature 1990, 346, 425 and references cited therein.
- (a) Kelly, T. A.; Jeanfavre, D. D.; McNeil, D. W.; Woska, J. R., Jr.; Reilly, P. L.; Mainolfi, E. A.; Kishimoto, K. M.; Nabozny, G. H.; Zinter, R.; Bormann,

- B.-J.; Rothlein, R. J. Immunol. 1999, 163, 5173; (b) Last-Barney, K.; Davidson, W.; Cardozo, M.; Frye, L. L.; Grygon, C. A.; Hopkins, J. L.; Jeanfavre, D. D.; Pav, S.; Qian, C.; Stevenson, J. M.; Tong, L.; Zindell, R.; Kelly, T. A. J. Am. Chem. Soc. 2001, 123, 5643.
- Kelly, T.; Fogal, S.; Giblin, P.; Haynes, N.; Jeanfavre, D.; Kishimoto, K.; Lai, G.; Lemieux, R.; Minolfi, E.; Panzenbeck, M.; Reilly, P.; Tweedie, D.; Woska, J.; Wu, J.-P. Presented at the 11th National Conference of the Inflammation Research Association, New York, USA, October 2002, S17.
- Wu, J.-P.; Kelly, T. A.; Lemieux, R.; Goldberg, D. R.; Emeigh, J. E.; Sorcek, R. J. Patent WO-2001007440, 2001.
- (a) Yee, N. K. Org. Lett. 2000, 2, 2781; (b) Frutos, R. P.;
  Stehle, S.; Nummy, L.; Yee, N. Tetrahedron: Asymmetry 2001, 12, 101.
- Appel, R.; Kleinstuck, R.; Ziehn, K. Chem. Ber. 1971, 104, 1335.
- The use of alternative dehydration conditions such as Burgess reagent, p-TsCl/Et<sub>3</sub>N or Ph<sub>3</sub>P·Br<sub>2</sub>/Et<sub>3</sub>N failed to give any of the desired 8.
- 8. Basha, A.; Lipton, M.; Weinreb, S. M. Tetrahedron Lett. 1977, 18, 4171.
- We noticed that the cyclization of 8 proceeded more efficiently in the presence of Ph<sub>3</sub>PO, and the mechanistic implications of this observation are currently under investigation.
- 10. Treatment of 5 with PBr<sub>3</sub>, POBr<sub>3</sub> and (COBr)<sub>2</sub> failed to give any vinyl bromide analogous to 3.
- For representative examples see: (a) Powers, J. C. J. Org. Chem. 1966, 31, 2627; (b) Russell, K.; Van Nivelt, C. E. J. Heterocycl. Chem. 1995, 32, 299; (c) Langley, B. W. J. Am. Chem. Soc. 1956, 2136; (d) Brennan, M. R.; Erickson, K. L.; Szmalc, F. S.; Tansey, M. J.; Thornton, J. M. Heterocycles 1986, 24, 2879; (e) Bendall, J. G.; Payne, A. N.; Screen, T. E. O.; Holmes, A. B. Chem. Commun. 1996, 11, 1067.
- 12. Barton, D. H. R.; Bashiares, G.; Fourrey, J.-L. *Tetrahedron* 1988, 44, 147.
- (a) Su, Z.; Paquette, L. A. J. Org. Chem. 1995, 60, 764;
  (b) Hirst, G. C.; Johnson, T. O.; Overman, L. E. J. Am. Chem. Soc. 1993, 115, 2992.
- Garcia Martinez, A.; Martinez Alvarez, R.; Garcia Fraile, A.; Subramanian, L. R.; Hanack, M. Synthesis 1986, 222.
- 15. Lee, K.; Wiemer, D. F. Tetrahedron Lett. 1993, 34, 2433.
- (a) Buon, C.; Bouysoou, P.; Coudert, G. *Tetrahedron Lett.* 1999, 40, 701;
  (b) Nicolau, K. C.; Sui, G.-Q.; Namoto, K.; Bernal, F. *Chem. Commun* 1998, 1757.
- 17. The TMSCl/NaI/water system generates HI in situ under mild conditions, and has been used for the synthesis of vinyl iodides via hydroiodination of alkynes, see: Kamiya, N.; Chikami, Y.; Ishii, Y. Synlett 1990, 675 and references cited therein for further details.
- 18. Immediate quench of the reaction after completion was required, otherwise continuous exposure of the product 3 to the reaction conditions resulted in various amounts of protodeiodination product 4.
- Wu, J.-P.; Emeigh, J. Presented at the 219th National Meeting of the American Chemical Society, Washington, DC, March 2000; ORGN-201.